# Cell–cell contacts confine public goods diffusion inside *Pseudomonas aeruginosa* clonal microcolonies

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The maintenance of cooperation in populations where public goods are equally accessible to all but inflict a fitness cost on individual producers is a long-standing puzzle of evolutionary biology. An example of such a scenario is the secretion of siderophores by bacteria into their environment to fetch soluble iron. In a planktonic culture, these molecules diffuse rapidly, such that the same concentration is experienced by all bacteria. However, on solid substrates, bacteria form dense and packed colonies that may alter the diffusion dynamics through cell-cell contact interactions. In Pseudomonas aeruginosa microcolonies growing on solid substrate, we found that the concentration of pyoverdine, a secreted iron chelator, is heterogeneous, with a maximum at the center of the colony. We quantitatively explain the formation of this gradient by local exchange between contacting cells rather than by global diffusion of pyoverdine. In addition, we show that this local trafficking modulates the growth rate of individual cells. Taken together, these data provide a physical basis that explains the stability of public goods production in packed colonies.

biofilm | evolution | noise | variability | ecology

ron is required for many enzymatic processes, and is therefore an essential nutrient. To overcome the low abundance of free iron in aerobic environments, bacteria, as well as other microorganisms, synthesize and secrete iron-chelating molecules called siderophores (1). Once released in the extracellular environment, siderophores diffuse and complex with iron. Because other bacteria can import them, siderophores can be regarded as public goods (2) and siderophore secretion is often cited as a model system for studying the stability of cooperation (3, 4). In this context, nonproducing mutants, which can enjoy the benefit of siderophores without bearing the cost of their production, have a selective advantage over the WT-producing strain and could, in principle, displace them on evolutionary time scales.

This argument usually assumes that public good molecules are readily available to all, as is the case in planktonic cultures, where molecules diffuse freely and rapidly between cells. In liquid conditions, mutants that do not produce siderophores have, in fact, been shown to outcompete WT strains (5-8). However, the limited spatial dispersal of public goods can challenge this picture and has been proposed as a general mechanism for explaining the maintenance of cooperation (8-12). When dispersal is limited, public good molecules tend to stay in the vicinity of the producing subpopulations, allowing them to benefit preferentially from their own production, and thus to balance the advantage of opportunistic nonproducing strains. In many ecological situations, bacteria form complex, tightly packed, and spatially structured colonies, such as biofilms, where public good dispersal may be severely reduced compared with planktonic cultures. This raises the questions of how public good molecules circulate between cells in these natural conditions and how cooperation is affected by the population structure (8, 13–15). Experimental realizations of limited public good dispersal were obtained by tuning the viscosity (16), the distance between colonies of producers and nonproducers (17), or the amount of cell attachment (18). Although the dispersal of public goods has been shown to influence the outcome of ecological competitions, direct measurements of public goods diffusion in structured population are still lacking.

Here, we address this question by taking advantage of the natural fluorescence of pyoverdine (Pvd), a siderophore produced by the human opportunistic pathogenic bacterium Pseudomonas aeruginosa (19). We monitored under a microscope the diffusion of Pvd in bacterial microcolonies forming monolayers. This minimal model for a spatially structured population of packed cells enables us to measure the distribution and spatiotemporal dynamics of siderophore concentrations precisely. Based on our observations at the single-cell level, we propose a model of local Pvd exchange and test its predictions on global, colony-wide observables. In particular, the model quantitatively explains the formation of a concentration gradient measured across the colony. We further show that the traffic of Pvd between cells has an impact on the growth rate at an individual level and discuss the ecological implications through observation-driven computer simulations of competition between producing and nonproducing strains.

#### Results

**Dynamics of Pvd Distribution Inside Clonal Microcolonies.** WT P. aeruginosa bacteria (PAO1) were inoculated at a low density on top of a nutritional gel sealed with a coverslip. We monitored colony growth by video microscopy thanks to a newly developed correlation imaging technique based on bright-field images taken at various focal positions (SI Appendix, Fig. S2). Every 3 min over the course of 7 h, we measured the endogenous fluorescence of free Pvd in each cell of a growing microcolony. We observed a strong variability in Pvd concentration across bacteria, as well as within the same bacteria at different times (Fig. 1A and SI Appendix, Movie S1). We reconstructed the genealogical trees of 10 microcolonies (Fig. 1B). Along each lineage, the fluorescence per pixel in cells fluctuated strongly in time (by about  $\pm 20\%$ ), with a correlation time of  $\sim$ 93 min (i.e., longer than the averaged division time of  $\sim$ 40 min). Although the average fluorescence per pixel in the colony increased linearly with time over the course of nine cell divisions (Fig. 1C), the distributions of cell fluorescence normalized by the average fluorescence of the colony remained invariant over time and across different colonies (Fig. 1D and SI Appendix, Fig. S3). The analysis of Pvd dynamics in individual cells reveals that the concentrations in both the cell and its neighbors appear to contribute to free Pvd dynamics (Fig. 1E and SI Appendix). Additionally, a gradient of Pvd is established between the border and the center of the colony, suggesting that Pvd dynamics are local.

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**Fig. 1.** Dynamics and variability of Pvd concentrations in *P. aeruginosa* (PAO1) microcolonies. (A) Time-lapse fluorescence images of a growing microcolony, with correlation images (*Upper*) and Pvd intrinsic fluorescence (*Lower*). (*B*) Genealogical tree of the colony. The level of Pvd in each cell *c* is color-coded along each lineage (for clarity, only a subset of the tree is displayed). (C) Dynamics of the mean level of Pvd  $\overline{c}$  in four different colonies. AU, arbitrary unit. (*D*) Distributions of Pvd concentrations in cells at various time points in a given microcolony (green), in a different microcolony (red), and in a medium supplemented with human transferrin (blue). (*Inset*) Distributions are normalized by the mean in the microcolony and the predicted distribution of our model (black). Error bars represent the SEM. (*E*) Dependence on the relative Pvd concentration in the cell  $x = c/\overline{c}$  and its neighbors  $x_{neigh}$  of the variation dx/dt. Color transparency indicates the uncertainty of measurements (light color means uncertain). The correlation between dx/dt and  $x_{neigh}$  is assessed with a *P* value less than 10<sup>-6</sup>.

**Model of Pvd Dynamics in Individual Cells.** The overall dynamics of Pvd are a product of a complex pathway involving synthesis, export, and import of the molecule. Pvd biosynthesis starts in the cytoplasm and ends in the periplasm (*SI Appendix*, Fig. S1). It involves four nonribosomal peptide synthetases and probably seven different other enzymes (20–22). Newly synthesized Pvd is stored in the bacterial periplasm (21) before secretion into the extracellular medium by the efflux system PvdRT-OpmQ (23). After iron chelation in the extracellular medium, ferri-Pvd is imported across the outer membrane by the ferri-Pvd outer membrane transporter FpvA (24) and iron is released from the siderophore in the periplasm (25). Free Pvd is then recycled into the extracellular medium by the efflux pump PvdRT-OpmQ (26, 27). Pvd biosynthesis is under positive feedback, regulated by a  $\sigma$ -factor PvdS

and its anti– $\sigma$ -factor FpvR, triggered when bacteria import ferri-Pvd (28, 29). Free Pvd is naturally fluorescent, whereas ferri-Pvd is not.

To model the spatial dynamics of Pvd inside microcolonies, we reasoned that the net effect of production, export, import, dilution (from cell division rate v), and degradation depends on the periplasmic concentration (dilution, export), on the concentration in the immediate vicinity of a bacterium (import), or on both (through feedbacks). Keeping only linear terms in the concentrations leads to a dynamical equation (*SI Appendix*):

$$\frac{dc}{dt} = \alpha - \lambda c + \kappa c_{\text{neigh}} + \text{noise},$$
[1]

where c and  $c_{\text{neigh}}$  stand for the concentration in the cell and its neighbors;  $\alpha$  is the basal production rate; and  $\lambda, \kappa > 0$  are

phenomenological parameters determined from the data (see below), reflecting the effective efflux of Pvd out of a cell and the influx from its neighbors. Note that because of the negative contribution from feedback and import,  $\lambda$  is expected to be smaller than v (Materials and Methods). The observed linear increase of the average fluorescence in the colony indicates a balance between efflux and influx,  $\kappa \approx \lambda$ . This balance defines an exchange mechanism formally equivalent to an effective diffusion process on the lattice formed by adjacent neighbors (SI Appendix). Within the mean-field approximation (SI Appendix), the above model predicts that local fluctuations in Pvd,  $c' = c - c_{\text{neigh}}$ , obey an Ornstein–Uhlenbeck process,  $dc'/dt = -\lambda c' + noise$ , meaning that the local fluctuations are damped proportionally to their amplitude with a measured rate of  $\lambda = 1.07 \pm 0.1 \cdot 10^{-2} \text{ min}^{-1}$  (Fig. 2A), which is independent of the average concentration in the colony (SI Appendix, Fig. S4). Consistent with our model, these local fluctuations are uniform within the colony (SI Appendix, Fig. S5) and spatially uncorrelated (SI Appendix, Fig. S7). The noise spectrum was observed to be white, and its intensity scaled quadratically with the average Pvd concentration in the colony:  $\langle noise(t) \cdot noise(t') \rangle = 2\Gamma \overline{c}^2 \delta(t-t')$ , with  $\Gamma = 2.19 \pm 10^{-2} \delta(t-t')$  $0.17 \cdot 10^{-4} \text{ min}^{-1}$  (Fig. 2B and SI Appendix, Figs. S8 and S9). With  $\lambda$  and  $\Gamma$  measured on data collected from 10 growing microcolonies of WT bacteria, we solved our model to predict the spatiotemporal dynamics of Pvd and its variability in packed microcolonies.

**Predictions of the Model for WT Microcolonies.** First, the experimental data confirmed the prediction of the model on the variability in the population. The variability is shown to scale with the average fluorescence in the colony, with a proportionality constant given by  $\sqrt{\Gamma/\lambda}$  (Fig. 3*A* and *SI Appendix*, Fig. S3). This linear relation, which also holds in a well-mixed condition (*SI Appendix*, Fig. S17), is unusual. In most descriptions of gene regulation, noise stems from fluctuations in the small number of molecules (30). The magnitude of these fluctuations is expected to increase not linearly but as the square root of the mean number of molecules. A linear scaling similar to ours has been reported for proteins produced in high copy numbers (31, 32), however, and has been attributed to sources of extrinsic noise. In the Pvd system, this high level of noise may be due to fluctuations in the number of efflux pumps and transporters.

Second, our model of local exchange explains the gradient of Pvd concentration observed away from the center of the colony (Fig. 3B). Cells on the edge of the colony have fewer neighbors to import from, and thus have lower Pvd levels. In turn, these cells contribute less Pvd to their direct neighbors. This effect is propagated toward the center, resulting in a gradient of Pvd across the colony. This gradient was calculated by solving Eq. 1 over the whole population (*SI Appendix*) and is in close agreement with the observed experimental gradient (Fig. 3B).

Third, adding the calculated spatial heterogeneities to the local fluctuations, we have been able to explain and predict the shape of the normalized Pvd distribution (Fig. 1D, Inset). In this framework, the variability is governed mainly by local exchanges rather than by fluctuations in production activity. Consistent with this interpretation, we found that the variability in production rate accounts for only 1% of the observed variance of the concentration of Pvd in cells expressing a PvdA fusion, the gene coding for the first enzyme involved in Pvd synthesis (Fig. 3D).

Finally, using Eq. 1, we computed the temporal correlations of spatial fluctuations (*SI Appendix*). This correlation decays exponentially with a characteristic time,  $\tau = 1/\lambda = 93$  min, in agreement with our measurements (Fig. 3*C*).

Predictions of the Model in Other Contexts. An important consequence of these local exchanges is that Pvd should be lost in the environment if bacteria do not interact physically by cell-cell contact. Accordingly, we observed that when bacteria are grown in stirred liquid conditions, Pvd is mainly found in the environment and the concentration in isolated bacteria is much lower than in packed bacteria (Fig. 4A and SI Appendix, Fig. S10). In contrast to passive diffusion, which leads to the fast release of Pvd in the extracellular environment in stirred conditions, the local exchanges between cells on solid substrate actively slow down the leak of Pvd out of the colony. We further challenged the predictions of our model by examining the effects of mutations in the Pvd pathway (SI Appendix, Fig. S1). In  $\Delta fpvA$  microcolonies, bacteria are not able to import Pvd from their neighbors and local trafficking of Pvd is prevented. In this strain, microcolonies cannot accumulate Pvd and no gradient is formed (SI Appendix, Movie S3). For a strain partially deficient in Pvd export, the model predicts that local fluctuations should be damped with a characteristic time longer than for a WT strain (SI Appendix). Consistent with this prediction, we measured a lower  $\lambda$  in the export-deficient  $\Delta pvdRTopmQ$  strain than in WT (Fig. 4B).

Local Exchange Affects Individual Fitness. The dynamics described above imply that iron uptake is mediated by local trafficking of Pvd when bacteria grow in packed colonies. To test whether this mechanism could enhance individual fitness, we measured the dependence of the individual growth rate on the history,  $c_n$ , of Pvd concentration in the neighboring cells (*SI Appendix*) through linear fits (SI Appendix, Fig. S11). On the addition of a competing iron chelator [human transferrin (Tsf)], growth became limited by iron. Individual growth rates then showed a positive dependence on the history of Pvd concentration in neighboring cells (Fig. 5A) and a moderate negative dependence on both the distance to the edge of the colony and the internal concentration (*SI Appendix*, Fig. S11). No dependence was observed in absence of Tsf (SI Appendix, Fig. S12). Interestingly, we observed that in iron-depleted conditions [succinate minimal medium (SMM) + Tsf], WT cells formed double layers at significantly earlier stages of microcolony development (SI Appendix, Fig. S12) than in

**Fig. 2.** Measurement of the model parameters. (A) Average change in relative Pvd concentration in a given cell is plotted (points with error bars) vs. its average relative concentration (relative to its nearest neighbors). (*Insets*) Observed linear correlation is explained by a simple model: Bacteria exchange Pvd with their neighbors in proportion to the difference of their concentrations. The proportionality coefficient  $\lambda$  is estimated by a linear fit to the data (continuous line). Error bars represent the SEM. (*B*) Temporal autocorrelation function of the noise for various values of the mean Pvd concentration in the colony,  $\bar{c}$ . This function can be approximated by  $\hat{A}e^{-t/\tau}$ , which we fit for each Pvd level. Because  $\tau$  is small, the noise autocorrelation



function can be approximated by a  $\delta$ -function:  $C_{\text{noise}}(t) = 2\hat{\Gamma}\delta(t)$ , with  $\hat{\Gamma} = \hat{A}\tau$ . (Inset)  $\hat{\Gamma}$  increases quadratically with the concentration:  $\hat{\Gamma} = \Gamma \overline{c}^2$  (dashed line: variation of  $\hat{\Gamma}$  as predicted from the estimation of  $\Gamma$  deduced from SI Appendix, Fig. S9).

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control conditions (SMM), extending the exchange mechanism to the vertical dimension at the onset of biofilm formation, and thus limiting Pvd leaks from the colony.

*In Silico* Competition of Producers vs. Nonproducers. Spatially structured environments are known to promote cooperation (8–10) and to maintain phenotypic diversity (33–35). To investigate the stability of producers in the context of mixed, packed populations on evolutionary time scales, we implemented numerical simulations



**Fig. 4.** Model predictions for a planktonic culture and for a mutant. (*A*) Model prediction for the internal Pvd concentration for well-mixed planktonic cultures vs. 2D growth on an agar plate. The distribution of Pvd in cells grown in liquid and solid conditions shows that cells from liquid cultures retain less Pvd in their periplasm, whereas the concentration in the environment is higher (*Inset*). This suggests higher leakage when cells are not in contact. Error bars represent the SEM (n = 10). (*B*) As expected by the model, a partial mutant of Pvd export ( $\Delta pvdRTopmQ$ ) exhibits a smaller Pvd exchange rate  $\lambda$  (evaluated as in Fig. 2A) than for the WT strain (PAO1). Error bars represent the SEM, and the number is indicated for each condition on the bar plot. The two samples are different with a *P* value of 0.04. The number of asterisks indicates the significance level (0.01<\*<0.05; \*\*\*<0.001).

Fig. 3. Model predictions for the WT strain PAO1. (A) SD of local fluctuations in Pvd is proportional to the mean fluorescence in the colony, as predicted by the model, with coefficient  $\sqrt{\Gamma/\lambda}$  (black line). Colors code for different colonies (N = 10). Each point is calculated from the distribution of fluorescence in a colony measured at a given time. (B) Pvd gradient is established from the edge to the center of the colony. The black points are obtained by pooling the data from 10 microcolonies in discrete bins according to their distances to the edge (white arrow). The red curve is the model prediction. Error bars represent the SEM. (C) Comparison between the measured (black) and the predicted (blue) temporal autocorrelation functions. The shaded areas show the uncertainty in parameter estimation. Error bars represent SEM (N = 10). The yellow arrow points to the same cell at different time points. Note that in A-C, the predicted curves are not the result of a fit. (D) Level of Pvd in individual bacteria is only weakly related to its production rate assessed by the reporter strain PvdA-YFP (AU).

using our model for Pvd dynamics. Producers and nonproducers were allowed to compete on a square lattice with a constant population size (*SI Appendix* and *SI Appendix*, Fig. S13). We varied the range of effective diffusion through exchanges and the cost of production (Fig. 5B and SI Appendix, Movie S4). When the rate of exchange, ( $\kappa \approx \lambda$ ), is faster than the growth rate,  $\nu$ , producers always go extinct (*SI Appendix*, Fig. S14). In contrast, cooperation was found to be dominant if  $\lambda < \nu$ . In the observed range of experimental costs (5, 6) (*SI Appendix*) and at the value of  $\lambda$  we measured, the population of producers was found to be stable. Hence, the effective slowing down of diffusion by active exchange between contacting cells limits the dispersal of the public goods to the producers and their kin, minimizing exchanges with non-producer mutants.

#### Discussion

In aquatic ecosystems, Pvd diffuses freely and is homogenized well by the agitation of the medium, such that the secreted molecules are shared by all. However, P. aeruginosa is also known to colonize many solid habitats. In rhizospheres, plants, and human lungs, colonies mature into biofilm in which Pvd diffusion is expected to be dominated by local interactions rather than by free diffusion. Our data show that competition for Pvd is governed by local exchange between close neighbors, which actively confines its diffusion within the microcolony. It challenges the general assumption that secreted molecules diffuse freely when bacteria are packed. In particular, if the bacteria exchange this public good at a rate that is slower than their growth rate, Pvd exchange benefits mostly the clonal patches of producers, ensuring the maintenance of cooperation. From an evolutionary perspective, our results emphasize how the spatial structure of the colony might constrain its evolution. On solid substrate, offspring remain physically close to each other after division, causing cells to interact more with cells of their own kin than with other cells. When exchanges are local, nonproducer cells may only benefit from the public good if they are in contact with producers, which only occurs at the boundary between patches of producing and nonproducing cells.

Although small molecules (<1kDa) diffuse at high rates in water, it is possible for microscale chemical gradients to form within biofilms. At the microscopic scale, our data provide a

Fig. 5. Local trafficking affects the individual fitness and stabilizes cooperation. (A) Mean relative growth rate of cells (black points) is plotted against the distance to the colony edge (normalized by the mean radius of the colony  $\overline{d}$ ) and the recent history of Pvd concentration in the cell's neighborhood  $c_n$  (normalized by  $\overline{c}$ ). The colored planes are bivariate linear fits to the data. No significant dependence is found when the level of iron is low (SMM) (SI Appendix, Fig. S11). By contrast, when iron is depleted (SMM + Tsf), the individual growth rates depend on both the position inside the colony and the Pvd history in the nearest neighbors.



(B) Phase diagram of the sustainability of cooperation in an *in silico* competition experiment between defectors and cooperators. The result of the simulation after 4,000 generations is shown as a function of the local exchange rate of Pvd  $\lambda$  and the cost of production. Cooperators are found to dominate in a wide range of values around the measured exchange rate  $\lambda/\nu \sim 0.6$ . The white line depicts the initial ratio. The black line depicts the measured value of  $\lambda$ , and the dashed lines depict the confidence interval. The time evolution of the proportion of nonproducers at the points (I, II and III) marked in B is shown in *SI Appendix*, Fig. S14.

complete description of how local exchange can generate gradients when bacteria are physically in contact with each other. At the cellular level, differential responses to these local environments [e.g., as gradients of gene expression (36) or position-dependent cell specialization (37)] may increase phenotypic variability and promote bacterial adaptation during infection (38). The gradients formed inside packed colonies may also be used by bacteria as a cue for positioning during biofilm development (39). In the case of *P. aeruginosa*, Pvd has been identified as a virulence factor in the development of infectious biofilms (28, 40–42) and confocal imaging has revealed that the Pvd pathway is not activated uniformly within biofilms (43, 44). Thus, elucidating the formation of these gradients at early stages of microcolony development may help us to understand better the mechanisms responsible for physiological heterogeneities in biofilms and their role in pathogenicity.

Contact-dependent interactions between bacteria are known to operate via tube-like secretion systems that bridge the periplasms of two neighboring bacteria (45–47). In contrast, Pvd must diffuse in the environment to chelate iron. Like Pvd, diffusible molecules have been shown to function over short distances (48). Restricted diffusion by cell compaction enhances access to public goods, which has been shown to provide a growth advantage (18, 49). However, the molecular mechanisms responsible for their confinement remain to be elucidated. Several hypotheses can be considered. If the secreted molecules have an affinity for the LPSs or other components of the bacterial cell wall, they would preferentially diffuse in the vicinity of the bacteria. Similarly, if a polymeric network fills the space between bacteria in packed colonies, the diffusible molecules could be jammed inside its mesh.

Beyond the social aspects of public goods diffusion in packed colonies, the local exchange between adjacent bacteria is reminiscent of paracrine signaling in eukaryotic tissues. Altogether, these aspects emphasize the growing recognition that bacterial colonies might share many more similarities with multicellular organisms than previously assumed.

#### **Materials and Methods**

**Culture Conditions.** Strains are described in *SI Appendix*, Table S1. They were inoculated in LB from glycerol stocks and grown overnight at 37 °C. The next day, the culture was washed and resuspended at a 100-fold dilution in fresh SMM for an overnight growth at 28 °C. SMM is a medium with a low iron concentration that activates Pvd synthesis (composition: 6 g-L<sup>-1</sup> K<sub>2</sub>HPO<sub>4</sub>, 3 g-L<sup>-1</sup> KH<sub>2</sub>PO<sub>4</sub>, 1 g-L<sup>-1</sup> [NH<sub>4</sub>]<sub>2</sub>SO<sub>4</sub>, 0.2 g-L<sup>-1</sup> MgSO<sub>4</sub> 7H<sub>2</sub>O, and 4 g-L<sup>-1</sup> sodium succinate, with the pH adjusted to 7.0 by adding NaOH). On the third day, bacteria were diluted 10<sup>4</sup>-fold and plated on a gel pad (1% agarose in SMM). The preparation was sealed with double-sided tape (Gene Frame; Fisher Scientific) to a glass coverslip. A duct was cut at the center of the pad to let oxygen diffuse into the gel. Microcolonies grew on the pad at 30 °C using a custommade temperature controller.

To create iron-depleted conditions, 5  $\mu M$  of human Tsf (Sigma) supplemented with 20 mM NaHCO\_3 was added to SMM and agarose gel. At this

concentration and after 24 h of growth, the total yield of a  $\Delta$ pvdA strain was reduced by a factor of 10 (*SI Appendix*, Fig. S15).

Image Acquisition and Analysis. The dynamics of Pvd concentration in each cell of a microcolony were monitored by time-lapse microscopy and epifluorescence. Colonies were observed with bright-field illumination using a magnification of 100× and 1.35-N.A. objective (Apo-ph1; Olympus) on an automated inverted microscope (IX81; Olympus) equipped with an Orca-R<sup>2</sup> CCD camera (Hamamatsu). The microscope, the camera, and the stage were actuated with a LabView interface (National Instruments). Correlation images were obtained by acquiring z-stacks of bright-field images. For each pixel, we computed the correlation of the intensity profile along the vertical direction with a reference kernel (SI Appendix, Fig. S2). Fluorescence excitation was achieved with a mercury vapor light source (EXFO X-Cite 120Q). Pvd was imaged with a 390(40)/475(64)-nm filter set using a dichroic beam-splitter at 405 nm (Semrock). Cell segmentation was performed on correlation images, and cell lineage was computed using a MATLAB (Math-Works) code developed in the laboratory of Michael Elowitz (California Institute of Technology) (51, 52). To compute the concentration of Pvd c in a cell, we subtracted the average fluorescence per pixel of the background from the average intensity per pixel in the given cell. In this study, an average intensity of 1,000 arbitrary units for a cell corresponds to ~500 molecules of Pvd.

**Significance of the Dependence of** dx/dt **on**  $x_{neigh}$ . To test whether dx/dt also depended on  $x_{neigh}$  and not only on x (Fig. 1*E*), we performed a multivariate fit of dx/dt with respect to x and  $x_{neigh}$ . The linear coefficient of  $x_{neigh}$  was found to be positive. We tested the significance of this positive correlation using a permutation test. We divided the whole dataset into 400 bins according to the value of x. Inside each of these bins, the values of  $x_{neigh}$  were randomly permuted, and the same multivariate regression was performed on the permuted data points. Repeated 1 million times, this procedure never yielded a linear coefficient for  $x_{neigh}$  as high as the one observed in the dataset that had not undergone any permutations, thus implying a P-value <10<sup>-6</sup>.

**Percentage of the Variability Explained by Production.** To assess the contribution of production in the variability of siderophore concentration, we measured the level of Pvd in single cells of a colony grown on agar pads and the level of PvdA (using the fusion reporter strain pvdA-YFP).We estimated the percentage of the total unexplained variability by subtracting the linear contribution of production measured by fitting Pvd against PvdA-YFP and computing the remaining variability (Fig. 3*D*). The fraction of unexplained variability ( $1 - R^2 = 99\%$ ) is the ratio of this remaining variability to the total variability. We also measured the correlation between the level of Pvd and the level of production in liquid conditions to avoid the variability due to cell–cell contacts (*Sl Appendix*, Fig. S16).

**Exchange Model.** The exchange model described by Eq. 1 can be rationalized in two ways. First, as shown in Fig. 1*E*, the time derivative of Pvd concentration correlates positively with the concentration in the cell's immediate neighborhood and negatively with the cell's own concentration. Eq. 1 is the simplest model one can write that accounts for this observation.

Alternatively, one can write a model accounting for all the events that may affect Pvd concentration and expand it to the lowest (linear) order. The

evolution of Pvd concentration in a single cell is governed by production (21), export (23, 27), import (53), dilution from cell division and possible degradation, and positive feedback (19, 20, 28):

$$dc/dt = \alpha + k_{\rm in}c_{\rm out} - k_{\rm out}c - \nu c + fc_{\rm out},$$
[2]

where  $c_{out}$  is the concentration outside the cell;  $\alpha$  is the production rate;  $k_{in}$  and  $k_{out}$  are the import and export constants, respectively;  $\nu$  accounts for dilution due to growth rate (and possible degradation); and f accounts for positive feedback depending on Pvd uptake [consistent with the observation of positive feedback loops involving the import channel (28)]. Because cells are tightly packed, it is not possible to estimate  $c_{out}$  directly from microscopy images. Instead, we assume it to be approximated by a linear combination of the concentration inside the cell c and the average concentration in its nearest neighbors (defined as a corona of one cell's width around the cell):  $c_{neigh}$ :  $c_{out} = ac_{neigh} + bc$ . Regrouping terms, we obtain Eq. 1, with  $\kappa = a(k_{in} + f)$  and  $\lambda = k_{out} + \nu - b(k_{in} + f)$ .

**Comparison Between Liquid and Solid Cultures.** To assess the importance of cell–cell contact, cells were inoculated at the same bacterial density in parallel

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on agar pads and in liquid. They were then allowed to grow for 9 h. Bacteria from the liquid culture were plated directly without dilution 5 min before fluorescence measurements (*SI Appendix*, Fig. S11).

**Simulation of Spatial Competition for Mixed Population.** We performed simulations of bacterial evolution on a 30  $\times$  30 square lattice with periodic boundary conditions. Each point was occupied by a bacterium, which was either a producer or a nonproducer. At each point, we simulated the evolution of Pvd, external iron, and internal iron using Euler's integration method with 10 time steps per generation. In cells, Pvd evolution is given by Eq. 1, where  $c_{\text{neigh}}$  was taken as the average in the four nearest neighbors.

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## Supplementary Material: Cell-cell contacts limit public goods dispersal inside *P. aeruginosa* microcolonies

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**Correlation imaging** Correlation images were reconstructed from a z-stack of microcolonies observed under bright field illumination. A z-stack is composed of a 32 frames taken by step of 200nm below and above the focal plane of a microcolony. For each pixel, the experimental intensity profile in the vertical direction is correlated with a reference profile defined by the derivative of a gaussian:  $-(z - z_0) \exp{-\frac{(z-z_0)^2}{2\sigma^2}}$ , where  $z_0$  is the position of the focus in the stack and  $\sigma$  corresponds to the typical size of a bacteria (700nm). The result of the correlation sets then the intensity of the pixel for the new correlation image.

From cell segmentation performed on correlation images, we obtained the individual mask for all the bacteria in the microcolony. If a pixel is outside all individual mask, it is then defined as a background pixel. To measure the fluorescence in each cell, we subtracted the average fluorescence of the background to the fluorescence measured on each mask. The concentration c is then given as the sum of the fluorescence over the pixels of the cell normalized by the area of the cell.

To calibrate the intracellular pyoverdine concentration, we spin-coated samples containing a known concentration of pyoverdine in polyvinyl alcohol and measured the intensity of fluorescence. In this study, an average intensity of 1000 arbitrary units for a cell corresponds to  $\sim 500$  molecules of pyoverdine.

**Exchange model.** For the  $\Delta f p v A$  strain, there is no uptake  $k_{in} = 0$  and no positive feedback f = 0. Hence, we expect  $\lambda^{\Delta f p v A} = k_{out} + \nu > \nu$  and  $\kappa^{\Delta f p v A} = 0$ . For the  $\Delta p v dRT/Q$  strain, which secretes less pyoverdine,  $k_{out}^{\Delta p v dRT/Q} < k_{out}^{WT}$ . Hence, we expect  $\lambda^{\Delta p v dRT/Q} < \lambda^{WT}$ .

**Mean colony behaviour.** The dynamics of the mean pyoverdine concentration in the microcolonies is obtained by averaging Eq. (1) of the main text over cells, yielding

$$d\bar{c}/dt = \alpha - \gamma \bar{c},\tag{1}$$

with  $\gamma = \lambda - \kappa$ , and where noise self-averages to zero. The bar  $\bar{c}$  denotes the colony mean. The solution to this equation is

$$\bar{c}(t) = \frac{\alpha}{\gamma} \left[ 1 - e^{-\gamma t} \right]$$

The observed linear dependence of  $\bar{c}(t)$  with t indicates that  $\gamma < 1/(400 \text{ min})$  (the total time of an experiment). To better estimate  $\gamma$ , we fitted the data for  $d\bar{c}/dt$  to a line:  $d\bar{c}/dt \approx \alpha [1 - \gamma t]$ , which gave  $\gamma \approx 10^{-3} \text{ min}^{-1}$ .

**Parameter estimation.** In order to estimate the remaining parameters of the model ( $\kappa$  and the noise), we used a mean-field approximation, which amounts to treating the average pyoverdine

concentration from the cell's neighbors  $c_{\text{neigh}}$ , as if it were the colony average. We thus write:

$$dc_{\text{neigh}}/dt = \alpha - \gamma c_{\text{neigh}}$$

as in Eq. (1). Subtracting this expression from Eq. (1) of the main text, we get

$$dc'/dt = -\lambda c' + \text{noise}$$

where  $c' = c - c_{\text{neigh}}$  is proportional to the fluorescence contrast between one cell and its neighbors. The equation for c' describes the dynamics of an Ornstein-Uhlenbeck process. To estimate  $\lambda$ , we binned the data points according to their value of c' (each point in a bin is then the fluorescence contrast of one cell at one time). For each bin, we estimated dc'/dt by perfoming a linear fit of  $c'(t + \delta t) - c'(t)$  versus  $\delta t$ , for  $\delta t$  ranging from 3 to 15 min. The result of this fit is displayed as a function of c' in Fig. 2A. A linear fit of dc'/dt against c' yields  $\lambda$ . The error on  $\lambda$  is the standard deviation obtained by repeating the procedure on the 10 experiments. We also checked that the value of  $\lambda$  did not depend on the absolute concentration c, by further binning data according to c, and repeating the procedure for each c-bin (Fig. S4).

We next computed the noise term:  $noise(t) = dc'/dt + \lambda c'$ . While  $\lambda$  did not depend on c it turns out that the noise depends strongly on the mean concentration of siderophores in the colony. Consequently we binned the noise data according to the fluorescence level (siderophore concentration c) and for each c-bin we looked at the temporal autocorrelation function

$$C_{\text{noise}}(t) = \langle \text{noise}(t_0) \cdot \text{noise}(t_0 + t) \rangle$$

see Fig. 2B. Here  $\langle ... \rangle$  denotes an ensemble average, *i.e.* an average over realisations of the noise and should not be confused with the colony mean (e.g.  $\bar{c}$ ). Each curve was fitted to an exponential:  $C_{\text{noise}}(t) = \hat{A}e^{-|t|/\tau}$ . Since  $\tau \sim 10$  min is smaller than all other typical times (e.g.  $\lambda^{-1}$  or the mean time between divisions  $\nu^{-1}$ ), we approximated this exponential by a Dirac delta-function. The magnitude of the noise is then equal to the integral of the autocorrelation

function:  $\int_{-\infty}^{\infty} C_{\text{noise}}(t) dt = 2\hat{A}\tau$ . As seen from the inset in Fig. 2B, the noise level depends quadratically on the average fluorescence (pyoverdine concentration at time  $t_0$ ) in the colony:

$$\langle \text{noise}(t_0) \cdot \text{noise}(t_0+t) \rangle = 2\hat{A}\tau\delta(t) = 2\Gamma\bar{c}(t_0)^2\delta(t).$$
 (2)

As a check, we repeated the analysis on the normalised concentration  $x = c/\bar{c}$ , whose dynamics is given in the adiabatic approximation  $((1/c)dc/dt \ll \lambda)$  by  $dx/dt = -\lambda x + \xi(t)$ , with  $\xi(t) =$ noise/ $\bar{c}$ . The adiabatic approximation was tested in all colonies and on average:  $\langle (1/c)dc/dt \rangle \sim$  $10^{-3} \min^{-1}$ . To validate the dynamical equation for the normalized variable x, we verified that the noise term  $\xi(t)$  was independent of the normalization c. We binned the data according to cand for each c-bin fitted the correlation function to an exponential:

$$C_{\xi}(t) = \langle \xi(t_0 + t)\xi(t_0) \rangle = A_0 e^{-|t|/\tau}$$

We found that the properties of the normalized noise  $C_{\xi}(t)$  were independent of c (Fig. S8, hence  $\hat{A} = A_0 \bar{c}(t_0)^2$ ), in particular its magnitude  $\Gamma = A_0 \tau$  is constant (inset). Thus we computed the mean value of  $\Gamma$  by averaging the values obtained from the data for each experiment (Fig. S9).

An Ornstein-Uhlenbeck process can be solved analytically to give the statistics of fluctuations at steady state:  $\langle c' \rangle = 0$  and  $\langle c'^2(t) \rangle = \Gamma \bar{c}^2 / \lambda$ . This prediction is compared to the experiments in Fig. 2B. All variances and standard deviation were estimated from data using an unbiased estimator with a 1/(N-1) normalization factor, where N is the sample size [1].

**Spatial model.** The mean concentration in the immediate neighborhood of a cell *i* can be written as:  $c_{i,\text{neigh}} = (1/n_i) \sum_{j \text{ neigh of } i} c_j$ , where  $n_i$  is the number of neighbors of cell *i*'s. Then we have:

$$dc_i/dt = \alpha + \kappa \left(\frac{1}{n_i} \sum_{j \text{ neigh of } i} c_j - c_i\right) - \gamma c_i + \text{noise} \equiv \alpha + \kappa [\Delta c]_i - \gamma c_i + \text{noise.}$$
(3)

which is a diffusion equation (with leakage) on the lattice formed by cell adjacency.  $\Delta$  is the lattice equivalent of the Laplace operator. Within the adiabatic approximation  $(1/c)(dc/dt) \ll \lambda$  (see above), we can write a similar equation for x:

$$dx_i/dt = \tilde{\alpha} + \kappa [\Delta x]_i - \tilde{\gamma} x_i + \xi(t), \tag{4}$$

where  $\xi(t)$  is defined as before,  $\tilde{\alpha} = \alpha/\bar{c}$ , and  $\tilde{\gamma} = \gamma + (1/\bar{c})d\bar{c}/dt$ .

To predict the average pyoverdine distribution in a colony, we calculated the steady state  $\langle x_i \rangle$  for each snapshot of a colony, by solving for  $d\langle x_i \rangle/dt = 0$  (*i.e.* Eq. (4) without the noise term) while fixing the average value of  $\langle x_i \rangle$  at the colony edge (< 1  $\mu$ m) from its measured value.  $\tilde{\alpha}$  is left as a free parameter enforcing  $\bar{x} = 1$ . The adjacency network was obtained by a Voronoi tessellation of the cell centers.  $\Delta$  was calculated from this adjacency network and then symmetrized. To plot Fig. 2C, the average values  $\langle x_i \rangle$  thus obtained were binned according to their distance to the colony edge. The average value of the relative contrast  $\langle x'_i \rangle \equiv [\Delta \langle x \rangle]_i$  is shown as a function of the distance to the edge in Fig. S5. The temporal autocorrelations of the fluctuations of  $x_i$  and  $x'_i = (\Delta x)_i$  are given by:

$$\langle \delta x(t_0) \delta x(t_0+t) \rangle = \Gamma \sum_a e^{-(\gamma + \kappa \mu_a)t} / (\gamma + \kappa \mu_a)$$
(5)

and

$$\langle \delta x'(t_0) \delta x'(t) \rangle = \Gamma \sum_a \mu_a^2 e^{-(\gamma + \kappa \mu_a)t} / (\gamma + \kappa \mu_a), \tag{6}$$

where  $\mu_a$  are the eigenvalues of  $\Delta$ . The temporal correlations of the local fluctuations  $x'_i$  are compared to the data in Fig. 2D. The spatial correlations of the fluctuations were calculated using

$$\langle \delta x_i(t) \delta x_j(t) \rangle = \Gamma \sum_{a,b} U_{ia} [\gamma + \kappa (\mu_a + \mu_b)/2]^{-1} U_{bj}$$
(7)

and

$$\langle \delta x'_i(t) \delta x'_j(t) \rangle = \Gamma \sum_{a,b} [\Delta U]_{ia} [\gamma + \kappa (\mu_a + \mu_b)/2]^{-1} [U\Delta]_{bj}, \tag{8}$$

where U is the matrix of eigenvectors of  $\Delta$ . The total spatial correlation function represented in Fig. S6,  $(\langle x_i - 1 \rangle (x_j - 1) \rangle$ , was calculated as the sum of the spatial heterogeneities,  $\overline{(\langle x_i \rangle - 1)(\langle x_j \rangle - 1)}$ ,and local fluctuations:  $\langle \delta x_i(t) \delta x_j(t) \rangle$ . Similarly, the theoretical distribution in the inset of Fig. 1D inset was estimated by adding Gaussian fluctuations of variance  $\langle \delta x_i(t) \delta x_j(t) \rangle$  to the distribution of the calculated  $\langle x_i \rangle$ . Fluctuations from the local background  $\langle \delta x'_i(t) \delta x'_j(t) \rangle$  are compared to data in Fig. S7.

History of Pvd concentration in the neighboring cells According to the exchange model, [PFe] is proportional to the concentration  $c_{\text{neigh}}$  of free Pvd in the neighbouring cells. The evolution of internal iron is given by uptake and dilution from cell division, where  $\nu = \log(2)/(\text{division time})$ is the growth rate:

$$d[\mathrm{Fe}]_{\mathrm{int}}/dt = \kappa[\mathrm{PFe}] - \nu[\mathrm{Fe}]_{\mathrm{int}}$$
(9)

Solving this equation yields  $[Fe]_{int}(t) = \int_{-\infty}^{t} \kappa c_{neigh}(t') exp[-\nu(t-t')]dt'$ . This quantity is called  $c_n$  in the main text.

Simulation of spatial competition for mixed population. As observed in the experiments, 10% of synthesized pyoverdine diffuses freely. In cells, pyoverdine evolution is given by Eq. (1) of the main text, where  $c_{\text{neigh}}$  was taken as the average in the four nearest neighbors. External iron was either made to diffuse homogeneously and rapidly across the population (consisently with the high diffusion constant of  $\text{Fe}^{2+} = 7.6 \cdot 10^{-6} \text{ cm}^2/\text{sec}$  in water [2]) or alternatively was assumed not to diffuse at all. Iron uptake was proportional to the concentration of the pyoverdine-iron complex, [PFe], calculated from the following equations:

• [P][Fe]/[PFe] =  $K_d^* = 500$  nM, where  $K_d^* = 2[\text{Tsf}]K_d(\text{Pvd})/K_d(\text{Tsf})$  is an effective dissociation constant calculated by assuming that all iron not complexed with pyoverdine is complexed with transferrin, which has two binding sites for iron ([Tsf] = 5  $\mu$ M,  $K_d(Pvd) = 0.5 \ 10^{-24} M$  [3] and  $K_d(Tsf) = 10^{-23} M$  [4])

- [PFe] + [P] = c<sub>out</sub> ~ c<sub>neigh</sub> + c, equivalent to setting a = b = 1. In the simulation we set, a=1, b=0, *i.e.* c<sub>out</sub> = c<sub>cneigh</sub> corresponds to the worst-case scenario for producers, as cells only benefit from their neighbors' production, and not directly from their own. Setting a=b=1, which might be more realistic, would give a further advantage of producers over non-producers.
- $[Fe] + [PFe] = [Fe]_{tot} = 2.5 \,\mu M$  [3].

The evolution of internal iron is given by Eq. 9. The pyoverdine production rate  $\alpha$  was chosen so that the average pyoverdine concentration in a monoclonal population of producers is  $\sim 2 \,\mu$ M (estimated from our data).  $\gamma = \lambda - \kappa$  was set to  $0.03\nu$ . Notice that since we want to simulate long-term evolution, we need to assume that pyoverdine concentration reaches steady state, which is why we assign a non-zero value to  $\gamma$ . The relative fitness f of each cell was calculated from the internal iron concentration as:

$$f = 2[Fe]_{int} / ([Fe]_{int} + [Fe]_{int}^{(0)}) - C,$$
(10)

where  $[Fe]_{int}^{(0)}$  is the wild type value (*i.e.* obtained from a monoclonal population of producers at steady-state), and *C* is the production cost (zero for non-producers and a tunable parameter for producers). The particular form of iron-dependence of the fitness was chosen so that:

- (a) it is zero when  $[Fe]_{int} = 0$
- (b) it saturates at large [Fe].

(c) as observed in the experiment (Fig 5.C): an 10% increase in  $[Fe]_{int}$  results in a 5% increase in growth rate. So, its relative dependency upon  $[Fe]_{int}$  around  $[Fe]_{int}^{(0)}$  is 50%, which set the exponent n of the general Hill function ( $f = 2[Fe]_{int}^n / ([Fe]_{int}^n + ([Fe]_{int}^{(0)})^n) - C$ ) equal to 1.

The fitness is here defined so that the expected number of offsprings after time  $\log(2)/\nu$  is  $2^{f}$ , where  $\nu$  is the wild type (producer) growth rate in a monoclonal colony at steady state, without the production cost. Consistenly, when  $[Fe]_{int} = [Fe]_{int}^{(0)}$  and there is no cost (C = 0), f = 1 and the expected number of offsprings is 2.

A continuous and uniform flux of external iron was injected at each point so that the amount of iron in all forms remains constant at 2.5  $\mu$ M. Competition was modeled as follows. At each generation and for each cell, we pick a random neighbor among the four nearest to compete with. The cell was replaced by its neighbor with probability max $[0, (2^{f_{\text{neigh}}} - 2^f)/(2^{f_{\text{neigh}}} + 2^f)]$ , where f is the fitness of the cell of interest, and  $f_{\text{neigh}}$  the fitness of its neighbor. We simulated the competition for 4,000 generations, starting with a population made of 50% non-producers and 50% producers randomly distributed on the lattice.

Estimation of the production cost from the literature. Since the number of offspring in our simulation, (*i.e* the growth rate, is given by the fitness defined in Eq. (10), the ratio  $\frac{\Delta\nu}{\nu_{\rm vet}} = \frac{\nu_{\rm mut} - \nu_{\rm wet}}{\nu_{\rm vet}}$  is a direct measurement of the relative cost of production that is depicted on the ordinate of the phase diagram (Fig. 4). Although we measured the parameter  $\lambda/\nu$  in our experiments, we cannot directly estimate the relative cost C of production because the benefits of ferri-pyoverdine uptake depends strongly on the spatial structure under the microscope. However, several studies have reported the outcome of competitions in liquid conditions between wild type strains (PAO1, natural isolates) and mutants that are defective for pyoverdine production ( $\Delta pvd$ , mutants derived from directed mutagenesis PAO6609, etc...) [5, 6, 7]. As in liquid conditions, every cells enjoys the same benefit, the ratio  $\frac{\Delta\nu}{\nu_{\rm wet}} = C$  only accounts for the production cost and can be easily estimated from the plots. The data reported in the literature are expressed as  $v = \frac{x_{\rm mut}(t_{\rm end})}{1-x_{\rm mut}(0)}$  [5],  $p = x_{wt}(t_{end})$  [6] or  $W = \frac{\ln(x_{\rm wt}(t_{\rm end})/x_{\rm wt}(0))}{\ln(x_{\rm mat}(t_{\rm end})/x_{\rm mut}(0)}$  [7], where  $x_{\rm mut}(t)$  and  $x_{\rm wt}(t)$  are the frequency of mutants and wild type cells, respectively. These

observables are related to our relative production cost by the following equations :  $C = \frac{\ln(v)}{\ln(2)N}$ ,  $C = \frac{\ln[(\frac{1}{p}-1)\frac{x_{\text{wtt}}(0)}{x_{\text{mutt}}(0)}]}{\ln(2)N}$  and  $C = \frac{1-W}{W}$  where N represents the number of generations elapsed from 0 to  $t_{end}$ . N is estimated from the density of cell inoculation, except for [6] where this number is given explicitly. In Table S2, we summarised the conditions of competitions and how the estimation of the relative production cost  $C = \frac{\Delta \nu}{\nu_{\text{wt}}}$  measured from data available in the literature.

P. aeruginosa strains	relevant characteristics	reference
PAO1	wild type strain	[8]
PAO1pvdRTopmQ	derivative of PAO1; $\Delta pvdRTopmQ$ , chromosomally integrated	[9]
PAO1 <i>fpvA</i>	derivative of PAO1; $\Delta f pvA$ , chromosomally integrated	[10]
PAO1 <i>pvdA</i>	derivative of PAO1; $\Delta pvdA$ , chromosomally integrated	[11]
pvdA-yfp	derivative of PAO1; pvdA-eyfp, chromosomally integrated	[12]

Table S 1: Strain Table.

Strains	Mix Ratio	Culture Medium	<b>Competition Time</b>	Measured Parameter	<b>Production Cost</b>
average over pairs [5]	1:1	CAA + Pvd	24h	v = 1.1	C = 2%
PAO1 vs PAO6609 [6]	1:1	$CAA + 100 \mu g.mL^{-1}Tsf$	6x24h	p = 0.34	C = 2.3%
PAO1 vs PAO6609 [7]	1:1	$CAA + 100mg.mL^{-1}Tsf$	72h	W = 0.96	C = 4.2%

Table S 2: Estimation of the relative production  $\cot C = \frac{\Delta \nu}{\nu_{wt}}$  measured from the data available in the literature. In [5], the cost is estimated from Fig. 5A (Pvd type I) with N=7. In [6], the cost is estimated from Fig. 3 (global and low r) with N=42. In [7], the cost is estimated from Fig. 4 (2.5g of Casamino acids) with N=7.



Fig. S 1: Diagram of Pvd pathway. Pyoverdine biosynthesis starts in the cytoplasm and ends in the periplasm. It involves four cytoplasmic non-ribosomal peptide synthetases (NRPS) and probably seven different other cytoplasmic and periplasmic enzymes [13, 14, 15]. Newly synthesized Pvd is stored in the bacterial periplasm [14] before secretion into the extracellular medium by the efflux system PvdRT-OpmQ [16]. After iron chelation in the extracellular medium, ferri-Pvd is transported across the outer membrane by the outer membrane transporter FpvA [17] and iron is released from the siderophore in *P. aeruginosa* periplasm [18]. Free-Pvd is then recycled into the extracellular medium by the efflux pump PvdRT-OpmQ [9, 19]. Pyoverdine biosynthesis is under a positive feedback regulated by two sigma factors PvdS and FpvI associated with their anti-sigma factor FpvR [20, 21]. When ferri-Pvd binds to FpvA, the outer membrane transporter interacts with FpvR. This interaction triggers the release of FpvI and PvdS that will activate the transcription of *fpvA* (light green) and all other genes involved in the pyoverdine pathway (light purple), respectively.



Fig. S 2: a) A z-stack of bright field images I(x, y, z) are taken around the focal position. The vertical step is set at 200nm. The bacteria appear white when the z-position of the objective is below the focal plane and black when the z-position of the objective is above the focal plane (z=0). b) Each pixel of the observation field is correlated in the z-direction with an analytical kernel Ker(z) that corresponds to the derivative of a gaussian function centered on the focal plane and with standard deviation  $\sigma_z = 700$ nm. c) Correlated image:  $I_c(x, y, z = 0) = \int dz' I(x, y, z') Ker(z - z')$ . d) Phase contrast image. Unlike phase-contrast images, the correlation images display regular fluctuations around the background level (compare (e) and (f)). In addition, the cell interior and membrane anticorrelate, enhancing the contrast of the images.



Fig. S 3: Standard deviation of heterogeneities in pyoverdine concentration c vs. mean pyoverdine concentration in colony  $\bar{c}$ . Black line is model prediction.



Fig. S 4:  $\lambda$  does not depend on absolute concentration. Same plot as Fig.2A in the main text, but here the data were pooled and color-coded according to the mean fluorescence (i.e.  $\bar{c}$ ) in the colony.  $\lambda$  can also be fitted for each colony separately. The value  $\lambda = 1.07 \pm 0.1 \cdot 10^{-2} \text{ min}^{-1}$  reported in the main text and used in this plot (black line) is the average of this fit over different colonies.



Fig. S 5: The normalised fluctuations  $c'/\bar{c} = (c - c_{\text{neigh}})/\bar{c}$  does not depend on the distance to the edge, except very close to it. Error bars represent SEM.



Fig. S 6: Spatial correlations of pyoverdine heterogeneities  $C(r) = \sum_{ij} c_i c_j \delta(|\vec{r_i} - \vec{r_j}| - r) / \sum_{ij} \delta(|\vec{r_i} - \vec{r_j}| - r) - \vec{c}^2$ . The prediction of the model (red) is obtained as the sum of two contributions: (*i*) the correlation functions of local fluctuations; (*ii*) the concentration heterogeneities due to boundary effects and obtained by solving the diffusion equation for a mean pyoverdine concentration set by the cells on the edge of the colony. Error bars represent SEM.



Fig. S 7: Same as Fig. S6, with  $c' = c - c_{\text{neigh}}$  instead of c. Fluctuation correlations are local. The negative lobe is an artifact of the fact that when i and j are neighbors,  $c_j$  contributes to  $c_{i,\text{neigh}}$ , and vice-versa, resulting in a negative correlation. Error bars represent SEM.



Fig. S 8: Same as Fig. 2B, but considering the normalised noise  $\xi(t) = \text{noise}(t)/\bar{c}$ . Temporal autocorrelation function of the normalized noise:  $C_{\xi}(t) = \langle \xi(t)\xi(t_0 + t) \rangle - \langle \xi(t) \rangle \langle \xi(t_0 + t) \rangle$ , for various values of the mean pyoverdine concentration in the colony,  $\bar{c}$ . As before, this function can be approximated by  $Ae^{-t/\tau}$ , which we fit for each pyoverdine level. We approximate  $C_{\xi}(t) = 2\Gamma\delta(t)$ , with  $\Gamma = A\tau$ . Inset:  $\Gamma$  shows no systematic dependence on  $\bar{c}$ . Dashed line: average  $\Gamma$  calculated from all 10 colonies (Fig. S9).



Fig. S 9: Same as Fig. S8, but plotted for each colony. For each plot,  $\Gamma$  is calculated from an exponential fit as before. The value  $\Gamma = 2.19 \pm 0.17 \cdot 10^{-4} \text{ min}^{-1}$  reported in the main text is an average over the plots for the different colonies.



Fig. S 10: (A) Protocol for comparing solid and liquid cultures. Liquid cultures are grown for 9h before being plated on solid agar pads. Correlation images of bacteria grown in solid (B) and liquid (C) conditions. (D) and (E), Pvd fluorescence of the bacteria shown respectively in (B) and (C).



Fig. S 11: (A) and (B) The mean relative growth rate  $(\nu = d(\ln L(t))/dt$ , where L(t) is the cell length) of cells (black points) is plotted against the distance to the colony edge (normalized by the mean radius of the colony d) and the recent history of pyoverdine concentration in the cell's neighborhood  $c_n$  (normalized by  $\bar{c}$ ). The colored planes are bivariate linear fits to the data. (A) No significant dependence is found when the level of iron is low (SMM). (B) By contrast, when iron is depleted (SMM+Tsf), the individual growth rates depend on both the position inside the colony and the pyoverdine history in the nearest neighbors. (C) and (D) Projection of the data onto the two planes. (C) The growth rate is positively correlated with the pyoverdine history in neighboring cells (slope:  $0.60 \pm 0.10$ ), (D) and negatively correlated with the distance to the colony edge (slope:  $-0.22 \pm 0.04$ ). (E,F,G) Linear, multi-variable fits were performed separately for each microcolony growing in the two conditions: SMM = control (orange, n=10)and SMM+tsf = iron depleted (green, n=11). The error bars represent the standard errors. (E) Linear coefficients  $\alpha$  and  $\beta$  were inferred from the fit  $\nu/\bar{\nu} = f_0 + \alpha d/\bar{d} + \beta c_n/\bar{c}$ , where  $c_n(t) = \int_{-\infty}^t dt' e^{-\nu(t-t')} c_{\text{neigh}}(t')$  is the pyoverdine history in the nearest neighbors reflecting the fraction of iron uptake by local trafficking (Eq. 9).  $\alpha$  quantifies the dependence of the growth rate upon distance to the edge reflecting perhaps iron depletion at the center, and  $\beta$ upon the history of pyoverdine in the neighborhood. (F) Linear coefficients  $\alpha'$  and  $\beta'$  and  $\gamma'$ were inferred from the fit  $\nu/\bar{\nu} = f_0 + \alpha' d/\bar{d} + \beta' c_n/\bar{c} + \gamma' c/\bar{c}$  in the same conditions. (G) Comparison of fits with two ( $\alpha$ ,  $\beta$ ) and three variables ( $\alpha'$ ,  $\beta'$ ,  $\gamma'$ ). In Fig. 4 we only depicted the 3D representations of the fits given in (A). Error bars represent SEM, N=10.



Fig. S 12: Area of the colony before the onset of double layer formation in the control medium (SMM) and in iron depleted conditions (SMM+Tsf). According to a permutation test, the area is significantly different in the two conditions ( $p < 2.10^{-16}$ ). Error bars represent SEM, N is indicated for each condition on the bar plot.



Fig. S 13: Schematic of a simulation cycle. A cycle is composed of 10 time steps. The simulation assumes synchronous divisions. During each cell cycle, pyoverdine dynamics is evaluated from the spatial model with the parameters measured in the experiments. The dynamics of iron uptake is proportional to pyoverdine uptake and iron consumption is assumed proportional to the dilution rate arising from growth. At the end of a cycle, individual fitness is evaluated as a function of iron concentration inside the cell and compared to the fitness evaluated in adjacent cells. Selection is carried out by enabling cells of highest fitness to reproduce at the expense of cells with lowest fitness. For clarity we picked up images from the simulation corresponding to an interval of 300 cycles.



Fig. S 14: The local exchanges stabilize cooperation. (A) Phase diagram of the sustainability of cooperation in an *in silico* competition experiment between defectors and cooperators. The result of the simulation after 4,000 generations is shown as a function of the local exchange rate of pyoverdine  $\lambda$ , and the cost of production. Cooperators are found to dominate in a wide range of values around the measured exchange rate  $\lambda/\nu \sim 0.6$ . The white line depicts the initial ratio. The black line depicts the measured value of  $\lambda$  and the dashed lines the confidence interval. (B) Time evolution of the proportion of non-producers at the points marked in A.



Fig. S 15: We compared the yield of a wild type strain (PAO1) with that of a productiondeficient strain ( $\Delta$ pvdA) in liquid culture. The culture medium was varied in order to modulate the level of iron availability: no Tsf (control medium: SMM), Tsf (iron depleted medium: SMM + 5 $\mu$ M Tsf), Tsf+Fe (SMM + 5 $\mu$ M Tsf + 100 $\mu$ M FeSO<sub>4</sub>), Tsf+Pvd (SMM + 5 $\mu$ M Tsf + 30 $\mu$ M Pvd). For every conditions, the medium is supplemented with 20mM of NaHCO<sub>3</sub>. Error bars represent SEM, N=3.



Fig. S 16: To avoid the variability due to cell-cell contacts, we measured the correlation between the level of pyoverdine and the level of production in liquid conditions . After plating bacteria on solid agar pad for observation under the microscope, we measured the signal of Pvd and YFP in individual cells of the fusion reporter strain pvdA-YFP). We estimated the percentage of the total unexplained variability by subtracting the linear contribution of production measured by fitting Pvd against PvdA-YFP and computing the remaining variability. The fraction of unexplained variability (1-R<sup>2</sup>=92%) is the ratio of this remaining variability to the total variability.



Fig. S 17: To measure the scaling of the variability in well-mixed environment, we cultivated bacteria in liquid conditions. We sampled them at different time to vary the average concentration. One point corresponds to a field of view with approximately 200 cells. In liquid conditions a linear scaling between the variability of Pvd concentration and the mean concentration is also observed. Although the model of local exchange is not relevant in this situation, the sources of the variability remain the same: noise in production, fluctuations in the number of transporters and efflux pumps. With the same sources of variability, we observed the same scaling. However since the effective exchange parameters ( $\lambda$ ,  $\kappa$ ,  $\Gamma$ , etc.) for bacteria growing in solution may be different than for bacteria growing on agar pads, we cannot use our measurements to make predictions for these different growth conditions.

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# **Supporting Information**

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Movie S1. Time-lapse recording of Pvd fluorescence in a WT microcolony (PAO1). Each frame is normalized by the mean intensity of the image. (*Inset*) Increase of the mean fluorescence during the experiment is displayed. The correlation images used to track individual cells are shown in *SI Appendix*, Movie S2.

### Movie S1

DN A C



Movie S2. Time-lapse recording using the correlation imaging (SI Appendix, Fig. S2) performed on a growing WT microcolony (PAO1).

Movie S2



Movie S3. (Left) Time-lapse recording of a import-deficient microcolony ( $\Delta f p v A$ ) in phase contrast. (Right) Fluorescence of Pvd of the same microcolony.

Movie S3

DNA C



**Movie S4.** Simulations of the maintenance of cooperation. (*Left*) Maps of producers (white) and nonproducers (black). (*Right*) For each map, Pvd is simulated according to our model. The three rows correspond to three different regions in the phase diagram shown in Fig. 4. The parameters of the first row are picked up at the upper right corner of this phase diagram, corresponding to a fast diffusion of Pvd and a high cost of synthesis. The parameters of the last row are picked up at the lower left corner of this phase diagram, corresponding to a slow diffusion of Pvd and a low cost of synthesis. The parameters of the middle row are picked up at the middle of this phase diagram and correspond to the intermediate case.

Movie S4

A Nd

## **Other Supporting Information Files**

SI Appendix (PDF)